RUDENKO, A.I.; KHARCHENKO, N.S., professor, zaveduyushchiy; ANGARSKAYA, M.A., dotaent, direktor.

Pharmacology of an Indian hemp species with leaves of St. John's wort type. Farm. i toks. 16 no.2:36-40 Mr-Ap '53. (MLRA 6:6)

1. Kafedra farmakologii Khar'kovskogo meditsinskogo instituta (for Ruden-ko and Kharchenko). 2. Khar'kovskiy nauchno-issledovatel'skiy khimiko-farmatsevticheskiy institut (for Rudenko and Angarskaya).

(Cannabis indica)

RUDENKO, A. I.

RUDENKO, A. I.: "Investigation of operational changes in the state of gasket material (as applied to the SYZ-NATI and DT-54 tractors)." Joint Academic Council, All-Union Sci Res Inst of the Mechanization of "griculture (VIM) and All-Union Sci Res Inst of the Electrification of Agriculture (VIESKh). Moscow, 1956. (DISSERTATION FOR THE DEGREE OF CANDIDATE IN TECHNICAL SCIENCE).

So.: Knizhnaya letopis', No. 25, 1956.

RUDENKO, A.I.

Effect of hypericum dogbane on diuresis. Farm. i toks. 19 no.1:43-45 Ja-F '56. (MLRA 9:5)

1. Kafedra farmakologii (zav.-prof. N.S. Kharchenko) Khar'kovskogo meditsinskogo instituta.
(DIURETICS

Apocynum cannabinum (Rus))

KHARCHENKO, N.S.,; RUDENKO, A.I.

Historical material on the Department of Pharmacology of the Kharkov Medical Institute (1805-1955) Farm. i toks. 19 no.1:54-56 Ja-F '56.

(PHARMACOLOGY, education hist. in Russia (Rus))

MALAKHOV, G.M., prof., doktor tekhn.nauk; BEZUKH, V.R., gornyy inzh.; RUDENKO, A.I., gornyy inzh.

Ways of increasing the efficiency of the complete mining of untouched blocks of ore and interchamber pillars. Gor. zhur. (MIRA 16:10) no.9:20-24 S '63.

1. Krivorozhskiy gornorudnyy institut.

RUDENKO, Aleksandr Ivanovich, kand. tekhn. nauk; ZAGORSKIY, G., red.;
PORHLEBKINA, M., tekhn. red.

[Save fuel and lubricants] Ekonom' toplivo i smazochnye materialy.
Moskva, Mosk. rabochii, 1962. 34 p.
(Tractors)

(Tractors)

THE ARREST WHICH STREET SHOUTH BEAUTHER THE FOUND TO SELECT STORE WATER SHOUTH THE SELECT SELECTION OF THE S

ARDASHEV, Gavriil Romanovich; BAZAROV, I.V.; MIKHAYLOV, I.N.; MORSHIN,
A.V.; POLOTSKIY, I.V.; HUDENKO, A.I.; SITNIKOV, A.P.; SPERANSOV, N.N.;
KRYUKOV, V.L., red.; DEYEVA, V.M., tekhn.red.

[Maintenance of tractors and agricultural machinery] Tekhnicheskoe obsluzhivanie traktorov i sel'skokhoziaistvennykh mashin. Moskva, Gos.izd-vo sel'khoz.lit-ry, 1961. 470 p.

(MIRA 14:4)

(Tractors--Maintenance and repair)
(Agricultural machinery--Maintenance and repair)

APPROVED FOR RELEASE: 06/20/2000 CIA-RDP86-00513R001445920001-5"

,是大学性的理论是是一种理解的主义。

RUDENKO, Aleksendr Ivanovich, kand.tekhn.nauk; KRYUKOV, V.L., red.; PROKOF'YEVA, L.N., tekhn.red.; BALLOD, A.I., tekhn.red.

[Manual on the use of petroleum products on collective farms] Spravochnik po neftekhoziaistvu v kolkhozakh i sovkhozakh. Moskva, Gos.izd-vo sel'khoz.lit-ry, 1960. 143 p.

(MIRA 13:11)

(MIMA I

(Petroleum product

3(7) S0Y/50-59-2-8/25

AUTHORS: Rudenko, A. I., Ponomarev, B. P.

TITLE: On the Development of Phenological Work (O razvitii fenolo-

gicheskikh rabot)

PERIODICAL: Meteorologiya i gidrologiya, 1959, Nr 2, pp 38 - 39 (USSR)

ABSTRACT: In November and December 1957 the first post-war conference

ference on Phenology jointly organized by the Geografiches-koye obshchestvo Soyuza SSR (Geographical Society of the USSR) and the Botanicheskiy institut i Zoologicheskiy institut Akademii nauk SSSR (Institute of Botary and Institute of Zoology of the Academy of Sciences, USSR). In this connection it is mentioned that phenology is no

on phenology took place in Leningrad: the All-Union Con-

longer to be considered a secondary discipline but an independent one. At present, the main task of phenology is the establishment of connections between seasonal natural phenomena and environmental conditions, primarily meteoro-

logical and hydrological factors. In the next few years

Card 1/2 it is planned to publish the "Phenological Characteristics

On the Development of Phenological Work

SOV/50-59-2-8/25

of the USSR". The conference laid down a system of positive measures for the further development of phenology in the USSR: coordination of the efforts of phenologists and competent authorities, development of uniform phenological observation and evaluation methods, publication of phenological yearbooks and a popular magazine of phenology and a series of compendia containing phenological research material. The conference also adopted measures for the expansion of the voluntary phenological network by using students, teachers, farmers, apiarists, etc. Some of these measures have already been realized. The slow progress in the development of a uniform phenological observation method is regretted.

Card 2/2

PHASE I BOOK EXPLOITATION SOV/2334		Sponsoring Agencies: USSR, Glavnoye upravienty gidrometeorologich-selvy ministerative sel'skogs knosygatus, Usrainakiy, natohno-isaledoysel'skiy gidrometeorologicheskiy institut, and Ukrainakaya akademiy sel'skoknozyaystvennykn nauk. Nesp. Rd.: G.P. Prikhot'ko: Rd.: Y.D. Pisassevakaya: Tech. Rd.	Braynins. This book is intended for and instructors in relate		Material of the Conference (Cont.) SQV/2384	tation 111	Hinter wheat in the Southern Agrams, pretion Moisture Reserves for the Moisture Providing Irrigation Buchinstiv 7 v. furnature as		Anticidates L. M. [Odessa Hydromet, Institute] Rainless and Wet. Periods in the Prichermomorakaya (Riack Sea) Sceppe		169		Devices and Methods	Consings, V.V. (State Mydrological Institute) Determining Evapora- tion from Drained and Mon-Drained Swamps by the Heat-Balance	202	\$15	And the fall follow in the Degeneration of Posatone and the Appearance of Phytophthora (Paraetto Pungt)	logy Council of the Ukrasa 243		Cor ac
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RUDENKO, A.I.; PONOMAREV, B.P.

Development of phenological research. Meteor. i gidrol no.2:
38-39 F *59.

(Phenology)

BRUK, H.M.; RUDENKO, A.I.

Effect of ginseng on basic processes of the higher nervous activity under experimental conditions. Fiziol. zhur. [Ukr] 4 no.6:834-836 N-D 158. (MPA 12:3)

1. Khar'kovskiy meditsinskiy institut, kafedra farmakologii. (GINNENG)

- 1. A. K. RUDENKO
- 2. USSR (600)
- 4. Agricultural Experiment Stations
- 7. Leading school farm. Dost sel'-khoz. no. 1. 1953.

9. Monthly List of Russian Accessions, Library of Congress, April 1953, Uncl.

Professor, A.K. (Bonetsk (obl.) 55, Universitetskava ul. d.25, kv.68)

fistory of the Borman-Levy bone-grafting emputation. Ortop.,

fravm. 1 protez. 25 nu.12:63-66 D 164.

1. it kacedry organizatali zdravookhraneniya i istorii nedittiny

(rektar - prof. A.M. Gunichkin). Submitted July 16, 1984.

Simulation of the state of the

SHABEL'NIKOV, G.P.; RUDENKO, A.M.

Controlling the piece-size distribution in breaking ore with long boreholes in the Salair Mine. Sbor. trud. VNIITSVETMET no.4:114-123 '59. (MIRA 16:8)

(Solar Ridge-Blasting)
(Boring)

Achievements of Dnieper River transportation workers. Rech.transp. (MIRA 10:12) 16 no.11:23-24 N 157. (Dnieper River-Inland water transportation)	DIMENSO A	. O , 17.	17).				en e
	Achi	evements of I	Onieper River '57. (Dnieper River	transportation	workers. (MIRA transport	Rech.transp. 10:12) ation)	

RUDENKO, A.M.; SPITSA A.I.; GROMOV, M.S.

Virusological characteristics of poliomyelitis in Dnepropetrovsk Province. Vop. virus. 7 no.2:240-241 Mr-Ap 162. (MIRA 15:5)

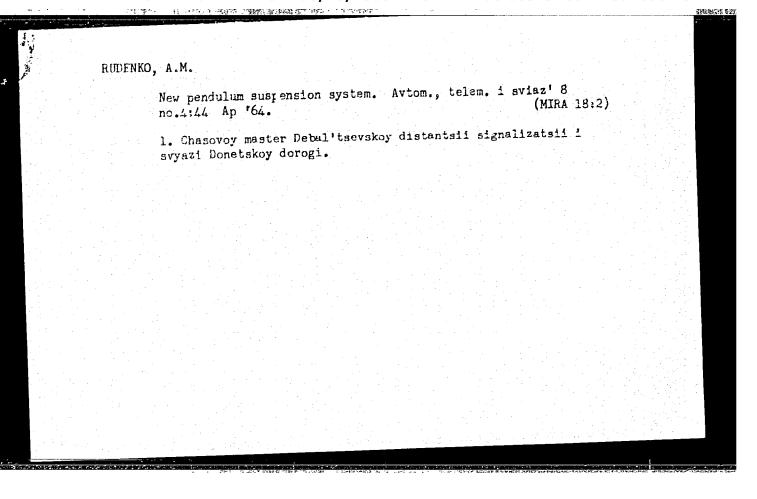
1. Dnepropetrovskiy institut epidemiologii, mikrobiologii i gigiyeny.
(DNEPROPETROVSK PROVINCE—POLIOMYELITIS)

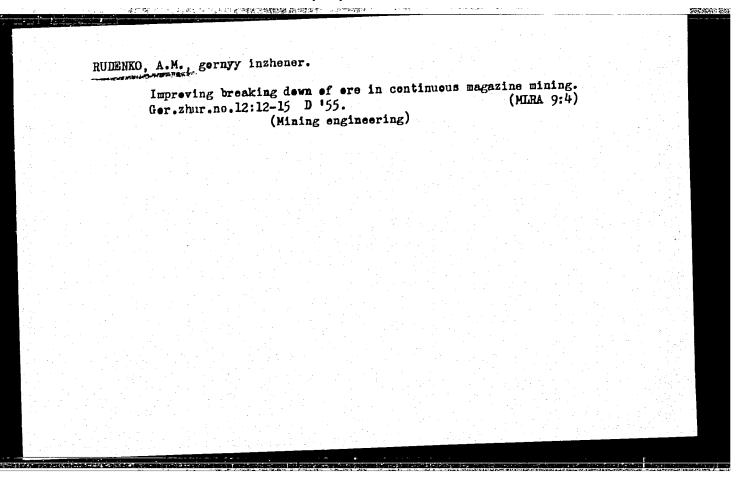
SHABEL'NIKOV, G.P.; LISOVSKIY, G.D.; STANKEVICH, I.M.; RUDENKO, A.M.; IEDYAYKIN, S.D.; ZEMLYANOV, V.P.

Testing a system of sublevel caving with breaking and drawing of the ore in inclined layers. Gor. zhur. no.6:23-24
Je 162. (MIRA 15:11)

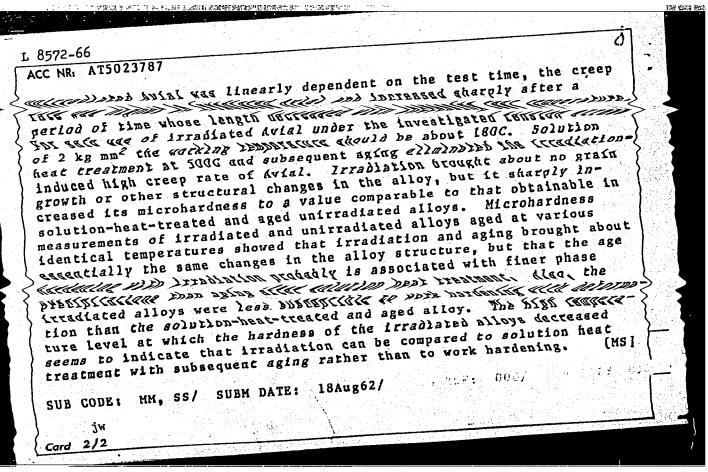
1. Vsesoyuznyy nauchno-issledovatel'skiy institut tsvetnykh metallov, Ust'-Kamenogorsk (for Shabel'nikov, Lisovskiy, Stankevich). 2. Salairskiy rudnik (for Rudenko, Ledyaykin, Zemlyanov).

(Salair region-Mining engineering)





	L 8572-66 EPF(n)+2/FWD(n)/FW4(1)/FW4	
	L 8572-66 EPF(n)-2/EWP(z)/EWA(h)/EWT(1)/EWT(m)/EWP(b)/EWA(d)/EWP(t) GG/MJW/JD AUTHOR:	
	Author: Petrov. P. A.; Batenin, I. V.; Rudenko, A. N.; Sherou P. 15.	_
	TITLE: Investigation of the properties of Avial irradiated in a	
	SOURCE :w Sove abob	
+	SOURCE: Soveshchaniye po probleme deystviye yadernykh izlucheniy na materialy. Moscow, 1960. Deystviye yadernykh izlucheniy na (The effect of nuclear radiation on materials); doklady soveshchaniya.	
- 1	12d-vo AN SSSR, 1962, 100-105.	
	TOPIC TAGS: Valuminum alloy, age hardenable alloy, neutron irradiated aluminum alloy resistance, neutron irradiation effect, /SAV-1	
1 '	ABSTRACT: Specimens of SAV-1 Avial, an aluminum-base alloy containing 0.011 Zn, 0.004 Ti, 0.48 Mg, and 0.001 Ni, were appealed to 2 hours for 2 hours	
	of 1019 n/cm2 cold acred irradiated at 800 with an integrated	
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i i	up to 260C. The test results showed that while the creep rate of.	j



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ACCESSION NR: AT5023796

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AUTHOR: Sharov, B. V.; Batenin, I. V.; Rudenko, A. N.

TITLE: X ray apparatus for structural study of radioactive materials

11 B+1

SOURCE: Soveshchaniye po probleme Deystviye yadernykh izlucheniy na materialy.

Moscow, 1960. Deystviye yadernykh izlucheniy na materialy (The effect of nuclear radiation on materials); doklady soveshchaniya. Moscow, Izd-vo AN SSSR, 1962, 180-183

TOPIC TAGS: x ray diffraction analysis, radioactive source, x ray apparatus

ABSTRACT: The chief difficulty involved in the use of a scintillation counter for recording soft x-ray quanta (Cu KA-radiation) in x-ray diffraction units is the elimination of the photomultiplier background. An improvement of the electronic part of the apparatus is proposed; it is established that an optimum supply voltage can be found for which the number of noise pulses having amplitudes equal to or greater than the amplitude of the pulses from x-ray quanta is negligibly small as compared to the intensity of the x-ray lines customarily recorded. The modification introduces a number of advantages: (1) Fewer parts are necessary to construct the unit (one-third as many radio tubes and resistances); (2) It is no longer necessary to convert the discriminators for coincidence

Card 1/2

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operation; (3) The efficiency of the	e apparatus is increased	by 50%; (4) Adjustment of	crystal
apparatus is improved because of	the convenient location	of the Mar(11) actilitization	
at the photomultiplier cathode. O	rig. art. nas: 4 figures		
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21 (9) AUTHORS:

Latenin, I. B., Rudenko, A. N.,

SOV/89-7-4-3/28

Sharov, B. V.

TITLE:

The Growth of Uranium Rods in an Aggressive Gaseous Medium

PERIODICAL:

Atomnaya energiya, 1959, Vol 7, Nr 4, pp 329-332 (USSR)

ABSTRACT:

The authors investigate rods made from technically pure uranium (diameter 2 to 4 mm, length up to 100 mm), the deformation texture of which had been removed by quenching. The extension of the rods was determined from the variation of the distances between the front surfaces of these rods, which had previously been polished until metallic luster was attained, or also from the variation of the distance between the marks previously made on the cylindrical surface. In some cases the extension was measured directly from the duration of the experiment by means of an indicator system. The gas pressure in the measuring apparatus could be varied between $10^{-2}\ \mathrm{mm}$ and atmospheric pressure. The temperature of the samples was controlled by means of a thermocouple. Heating of the samples with 4 mm diameter at atmospheric pressure led to a change of length. The first 3 diagrams illustrate the dependence of the growth of the rods on pressure at the temperatures of the

Card 1/4

The Growth of Uranium Rods in an Aggressive Gaseous Medium

SOV/89-7-4-3/28

 α -, β -, and γ -phase. The fourth diagram gives data concerning the dependence of the rate of increase of the rods on their diameter. Conditions otherwise remaining the same, samples, which have a thin oxide film on their surface, increase in length more rapidly than such as have a pure surface. The rate of increase at 500° C somewhat exceeds the rate of increase of the quenched rods. At normal pressure and at temperatures corresponding to the \$- and 7-phase, the samples extend when heated in nitrogen. Experiments carried out at atmospheric pressure in carbon monoxide gas prove the increase of the size of the rods at temperatures corresponding to the T-, B-, and α -phase. The density of the metal after the increase of volume is practically the same as the initial density. The increase in rod volume at the temperatures of the 6- and &-phase does not change the density of the sample. The surface of a uranium rod which has grown in volume when heated in air has a cubic face-centered lattice with the parameter 5.31 %. This lattice corresponds to the structure of UO2. In conclusion, the volume increase of copper wires is dealt with. A copper

Card 2/4

The Growth of Uranium Rods in an Aggressive Gaseous Medium

SOV/89-7-4-3/28

wire having a diameter of from 0.5 to 1 mm increases in volume by several per cent when heated up to 9000 in air within 30 minutes. With conditions otherwise being equal, the rate at which these wires increase in volume is inversely proportional to their diameter. Also the state of the wire surface exerts an influence on the increase in its volume. Finally, a possible mechanism for the volume increase of uranium rods is dealt with: Oxygen diffuses into the heated uranium rod, so that a film of the lowest oxides (UO + UO2) is formed. Oxidation is irregular and independent of crystallographical directions. Thus, it is possible to observe a colored mosaic on the electropolished uranium surface. The planes (020) have the highest degree of oxidizability, and the planes (002) the lowest. In the course of time also the lowest oxides oxidize with progressing oxidation processes. The increase in the rod volume is caused by oxygen which diffuses into the layer and oxidizes the lowest oxide. The oxygen exercises its most intensive effect with respect to the volume increase of uranium rods if the conditions corresponding to the production of the lowest oxides exist. There are 5 figures, 2 tables, and 1 Soviet

Card 3/4

The Growth of Uranium Rods in an Aggressive Gaseous SOV/89-7-4-3/28

Medium

reference.

SUBMITTED:

February 13, 1959

Card 4/4

PHASE I BOOK EXPLOITATION

Sov/6176

Konobeyevskiy, S. T., Corresponding Member, Academy of Sciences
USSR, Resp. 21.

Deystvive vademykh izlucheniv na materialy (The Effect of
Nuclear Radiation on Materials). Moscow, Izd-vo AN SSSR,
Nuclear Radiation on Materials). Moscow, Izd-vo AN SSSR,
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Occupancy of State of SSSR,
Occupancy of SSSR,
Otdeleniye tekhnicheskikh nauk;
Otdeleniye fiziko-matematicheskikh nauk;
Cheskikh nauk; Otdeleniye fiziko-matematicheskikh nauk;
Adasinskiy; Editorial Board: P. L. Gruzin, G. V. Kurdyumov,
Adasinskiy; Editorial Board: P. L. Gruzin, G. V. Kurdyumov,
B. M. Levitskiy, V. S. Lyashenko (Deceased), Yu. A. Martynuk,
B. M. Levitskiy, V. S. Lyashenko (Deceased), Yu. A. Martynuk,
House: M. G. Makarenko; Tech. Eds: T. V. Polyakova and
I. N. Dorokhina.

Card 1/14

90

307/6176

The Effect of Nuclear Radiation (Cont.)

PURPOSE: This book is intended for personnel concerned with nuclear materials.

COVERAGE: This is a collection of papers presented at the Moscow Conference on the Effect of Nuclear Radiation on Materials, held December 6-10, 1960. The material reflects certain trends in the work being conducted in the Soviet scientific research orginization. Some of the papers are devoted to the experimental study of the effect of neutron irradiation on reactor materials (steel, ferrous alloys, invadiation on reactor materials (steel, ferrous alloys, invadiation on reactor materials (steel, ferrous alloys, invadiation, avial, graphite, and nichromes). Others deal molybdenum, avial, graphite, and nichromes) others deal with the theory of neutron irradiation effects (physico-chemical transformations, relaxation of internal stresses, internal friction) and changes in the structure and properinternal friction; and changes in the structure and properities of various crystals. Special attention is given to the effect of intense Y-radiation on the electrical, magnetic, and optical properties of metals, dielectrics, and semiconductors.

Card 2/14

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The Effect of Nuclear Radiation (Cont.)	6176	
Lyashenko, V. S. (Deceased), and Sh. Sh. Ibragimov. Effect of Neutron Field on Structure and Properties of Steels of Neutron Field on Structure and Properties of Steels The specimens were irradiated in the fast reactor BR-5 with a neutron flux of 1.9 · 10 ² on/cm ² at temperatures from 150 to 220 ² [0?].	74	
Pronman, I. M., V. A. Shalashov, and A. Kn. Breger. Decomposition of Carbide Phase in Iron-Carbide Alloys and Phase Transformation in White Cast Iron Under Nuclear Irradiation	~~	
Petrov, P., A., I. V. Batenin, A. N. Rudenko, and B. V. Sharov Investigation of Properties of Avial Subjected to Nuclear Radiation in a Reactor	100	
Platonov, P. A. Stress Relaxation in Metals Under Platonov, P. A. Stress Relaxation in Metals Under Platonov, P. A. Stress Relaxation in Metals Under Platonov, P. A. Stress Relaxation in Metals Under	106	
Defects Specimens were irradiated at =150°C by fast neutron Specimens were irradiated at =150°C by fast neutron fluxes (E>1 mev) of 2.101° and 4.101° n/cm² in the RFT Reactor.		
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•	The Effect of Nuclear Radiation (Cont.)		
•	Batenin, I. V. V. A. Il'ina, V. K. Kritskaya, G. V. Kurdyumov and B. V. Sharoy. Investigation of the Effect of Neutron and B. V. Sharoy. Investigation of the Effect of Neutron and Properties of Investigation on Thin Crystalline Structure and Properties of	, 160	
	Metals and Alloys Annealed specimens (copper at 400°; iron and iron-nickel at 600°; iron-chromium and iron-tungsten at 650°; and at 600°; iron-chromium and iron-tungsten at 650°; and chromium at 900°) were irradiated with neutron fluxes of chromium at 900°) were irradiated with neutron fluxes of chromium at 900°) at a temperature not exceeding 80°[C?].		
	Karpukhin, V. I., and V. A. Nikolayenko. Remote Controlled Installation for X-Ray Diffraction Analysis of Radioactive Specimens	168	
	Levitskiy, B. M., and Yu. A. Martynyuk. Installation for X-Ray Examination of Highly Active Specimens	173	
	Sharov, B. V.) I. V. Batenin, and A. N. Rudenko. X-Ray Unit for Structural Investigation of Radioactive Materials	180	
	Card 8/14		

21(1), 18(7)

sov/89-6-5-11/33

AUTHORS:

Batenin, I. V., Rudenko, A. N., Sharov, E. V.

TITLE:

Dilatometric Investigation of Rolled Uranium Rods (Dilatometricheskiye issledovaniya prokatannykh sterzhney urana)

PERIODICAL:

Atomnaya energiya, 1959, Vol 6, Nr 5, pp 565-567 (USSR)

ABSTRACT:

Technically pure uranium which was rolled into rods of 4 mm diameter at ~300°C and at high pressure, was investigated in a vacuum dilatometer, and the course of the dilatometric curves for the first thermal cycle was found to be anomalous. After heating up to 525°C and subsequent cooling the dilatometric curves correspond to the known curves for rods with saturated axial structure [010]. An anomalous course of the curves is found in the case of cooling down also if heating during the first cycle ranged between 200 and 500°C. If the uranium is heated up to ~180°C, the curve for cooling coincides with that for heating, whereas in the case of heating up to more than 180°C the curves do not coincide. Heating up to temperatures of from 250° to 520°C shortens the rods. The rods shortened by the first thermal cycle are characterized by a noticeable shift of the inversion point (up to ~400°C). In the case of cooling down to ~600°C the inversion point is again shifted down to a temperature of ~ 200°C. If, during

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Dilatometric Investigation of Rolled Uranium Rods SOV/89-6-5-11/33

the second thermal cycle, the temperatures which correspond to inversion point are not exceeded, the curve for cooling practically coincides with that for heating. The results obtained by measurements are shown by diagrams; the measuring methods employed are not dealt with in detail. The anomalous course taken by the dilatemetric curves during the first heating of an uranium rod may possibly be connected with the diffusion of the impurities still existing in the technically pure uranium. It is possible that the said anomaly does not occur in the case of uranium of an especially high degree of purity. There are 3 figures and 2 references, 1 of which is Soviet.

SUBMITTED: November 25, 1958

Card 2/2

S/079/62/032/002/003/011 D227/D303

Rudenko, A.P.

TITLE:

AUTHOR:

Role of complex-forming additives in the synthesis of phthal ocyanines III. Mechanism and direction of catalytic actio vity of complex-forming additives in the synthesis of Fephthalocyanine

PERIODICAL:

Zhurnal obshchey khimii, v. 32, no. 2, 1962, 531-538

TEXT: A study of the effects of additions of phosporic acid, ammonium phosphate, molybdenum and tungsten trioxides, molybdic and tungstic acids, molybdo- and tungsto-phosphoric acids and their ammonium salts and chromium oxide on the rate of formation of Fe-phthalocyanine from iron and phthalimide in the presence of urea. From observations it followed that there was a definite concentration ratio of the additive to the reagents which together with the degree of stability of the intermediate products determined positive or negative catalytic effect of such additives. The investigations were based on the assumption that in the reaction ammonium

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Role of complex-forming ...

like salts are formed between the amide groups and acidic additives and that the catalytic activity of these additives is exerted through such intermediate ammonium-like compounds. From the considerations it followed that the acidity of the additives had no direct connection with their catalytic activity, but that the latter depended on the stability of the ammonium-like compounds or even more on the relative stability of the original acids and their ammonium salts. The lower was the stability of these compounds the easier were they decomposed and the higher was the positive catalytic effect of the corresponding additives. In investigating the direction of the catalytic activity of the additives the

author considers three possible cases: 11) Formation of stable compound; 2) formation of unstable compounds and 3) formation of 'in-between' com-From his analysis of these cases it follows that the change from positive to negative catalytic effect resulting from the increased concentration of the additives and decreased concentration of reactants, as compared with the equilibrium value, has no effect on the mechanism of the catalytic activity. The equilibrium is reached for different quantities of additive according to the stability of the intermediates. In the

Card 2/3

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Role of complex-forming ...

case of stable intermediates this equilibrium occurs when only small amounts of additive are present and in the case of unstable intermediates when very large amounts are present. There was no difference between the mechanisms of positive and negative catalytic activity of the additives and the nature of the intermediate compounds was exactly the same. There are 3 figures and 18 references, 15 Soviet-bloc and 3 non-Soviet obloc. The references to the English-language publications read as follows: F. Gager, Ind. Eng. Chem., 25, 1122(1933); R. Hansford, Ind. Eng. Chem., 39, 849, (1947); C. Thomas, Ind. Eng. Chem., 41, 2564 (1949).

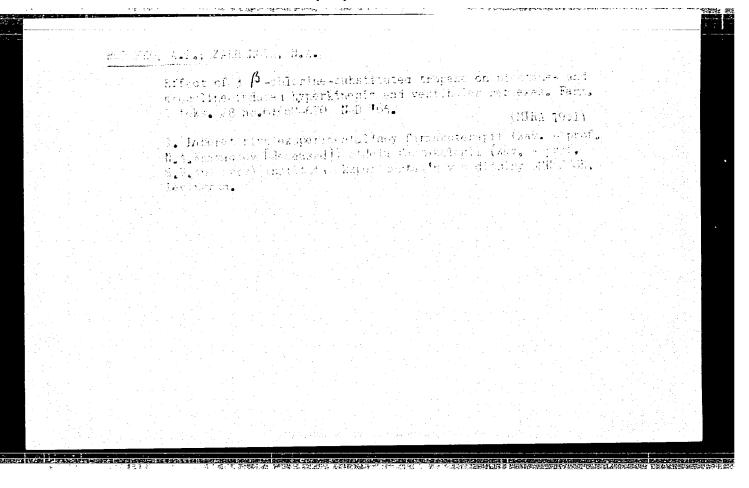
ASSOCIATION: Moskovskiy gosudarstvennyy universitet in. M.V. Lomonosova

(Moscow State University in M.V. Lemonsov)

SUBMITTED:

March 11, 1961

Card 3/3



ACC NR. AP6034265 (N) SOURCE CODE: UR/0390/66/029/005/0609/0611

AUTHOR: Rudenko, A. P.; Zakharova, N. A.

ORG: Division of Pharmacology / Head-Active number AMN SSSR S. V. Anichkov/, Institute of Experimental Hedicine, AMN SSSR, Leningrad (Otdel farmakologii Instituta eksperimental noy meditsiny AMN SSSR)

TITLE: Toxicity of certain tropage derivatives and their effect on hyperkinesia

SOURCE: Farmakologiya i toksikologiya, v. 29, no. 5, 1966, 609-611.

TOPIC TAGS: drug effect, tropane, tropane derivative, hyperkinesia, CHE-ELECT, N cholinolytic effect, stereoisomer, loxicity, central nerous system

ABSTRACT: The toxicity and central nervous system effects of the tropane derivatives shown in the figure were investigated. Table 1 shows the relative toxicity of the three compounds tested. The effect of these 3-substituted tropanes on hyperkinesia in rats were compared with those of corresponding stereoisomers. Only 3 alpha-chloronortropane produced central N-cholinolytic effects. Orig. art. has: 1 figure and 2 tables.

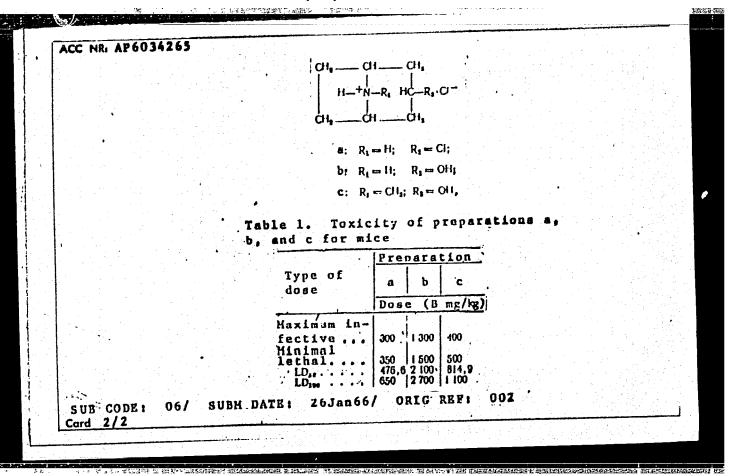
[W.A. 50]

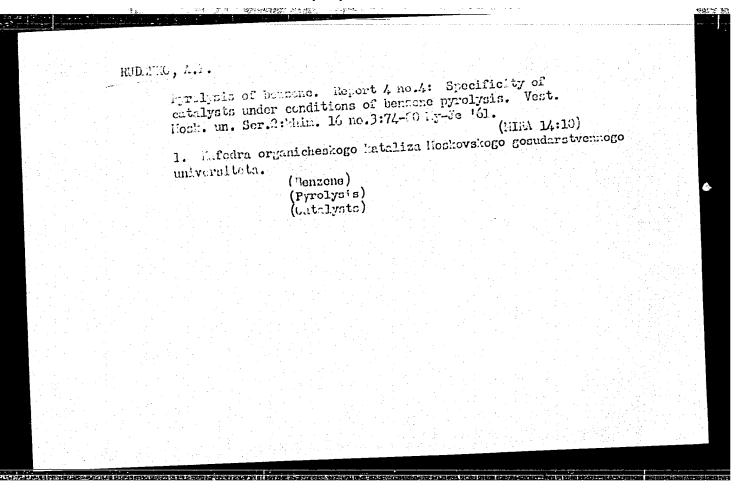
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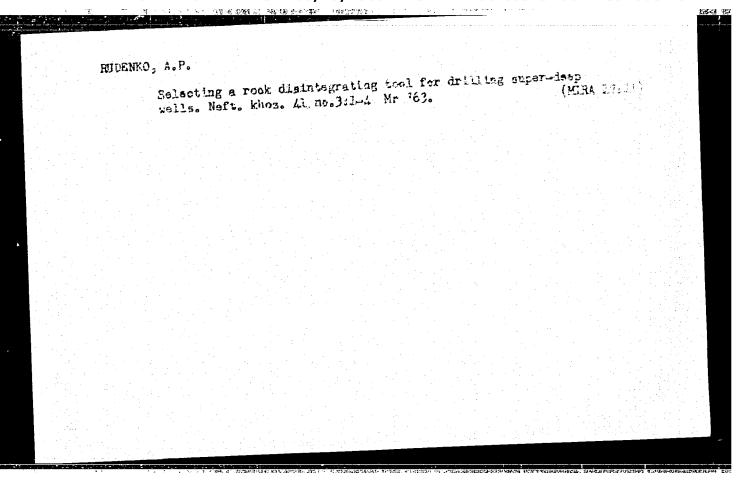
Card 1/2

APPROVED FOR RELEASE: 06/20/2000 CIA-RDP

CIA-RDP86-00513R001445920001-5"







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Self-developing catalytic systems. Dokl. AN SSSR 159 no.6:
1374-1377 D '64 (MIRA 18:1)

1. Moskovskiy gosudarstvennyy universitet. Predstavleno akademikom A.A. Balandinym.

RUDENKO, A.P.; BALANDIN, A.A.

Dehydrocondensation of methane with the formation of a coalyielding substance. Kin.i kat. 2 no.4:529-533 JI-ag '61.
(MIRA 14:10)

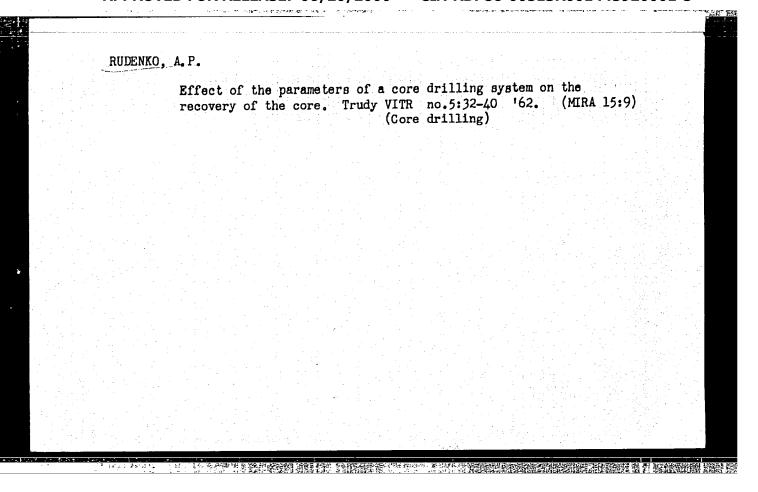
1. Moskovskiy gosudarstvennyy universitet imeni M.V.Lomonosova.
(Methane) (Condensation products (Chemistry))

BALANDIN, A.A.; SPITSYN, V.I.; RUDENKO, A.P.; DOBROSEL'SKAYA, N.P.; MIKHAYLENKO, I.Ye.; PIROGOVA, G.I.; GLAZUNOV, P.Ya.

Apparatus for studying heterogeneous catalysis at high temperature using radioactive catalysts and ionizing radiations. Kin.i kat. 2 no.4:626-632 JI-Ag '61. (MIRA 14:10)

1. Institut fizicheskoy khimii AN SSSR i Moskovskiy gosudarstvennyy universitet imeni M.V. Lomonosova. (Catalysis)

		i	Role of discolut	alkali m ion of d	etal hy Lumond.	droxide Dokl.	s and AN SSSI	carbon R 163 i	ates in	the 69-11	oxidizin 72 Ag ' (MIRA	65 .	
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s/062/60/000/011/003/016 B013/B078

AUTHORS:

Stegner, G., (?udenko, A. P., Balandin, A. A.

TITLE:

Carbon Formation in the Decomposition of Isopropyl Alcohol, n-Hexyl Alcohol, and Cyclohexanol on the

Copper - Silica Gel Catalyst

PERIODICAL:

Izvestiya Akademii nauk SSSR. Otdeleniye khimicheskikh

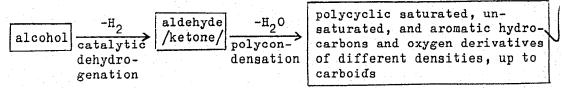
nauk, 1960, No. 11, pp. 1930 - 1937

TEXT: A study has been made of the mechanisms of carbon formation in the decomposition of isopropyl and n-hexyl alcohols, as well as of cyclohexanol in the temperature range of 200 - 950°C, proceeding in the same manner as with ethyl alcohol (Refs.1,2). The experiments were conducted in a continuous system for heterogeneous catalytic studies at atmospheric pressure and a volume velocity of 2.25 h⁻¹. Fig.1 shows the temperature dependence of the carbon formation rate in the decomposition of the above-mentioned substances. The presence of three mechanisms can be inferred from the course of the curves. Diagrams are suggested for the three mechanisms: the low-temperature mechanism in the temperature

Card 1/5

Carbon Formation in the Decomposition of S/062/60/000/011/003/016
Isopropyl Alcohol, n-Hexyl Alcohol, and B013/B078
Cyclohexanol on the Copper - Silica Gel Catalyst

range of 200° - 600° C acts, like a pure polycondensation of aldehydes and ketones, according to the following scheme:



This scheme is applicable to all alcohols concerned, including ethyl alcohol. The intermediate mechanism in the temperature range of 600° - 750° C is a polycondensation of products of the catalytic dehydration of alcohols (propylene, hexylene, cyclohexene). Possibly, a polycondensation of ethylene takes place likewise under these conditions, but since the rate of this process is too low, it occurs only at higher temperatures, in the course of the high-temperature mechanism. The following scheme is offered for the intermediate mechanism:

Card 2/5

"APPROVED FOR RELEASE: 06/20/2000

Carbon Formation in the Decomposition of S/OISOPROPYL Alcohol, n-Hexyl Alcohol, and BOCyclohexanol on the Copper - Silica Gel Catalyst

S/062/60/000/011/003/016 B013/B078

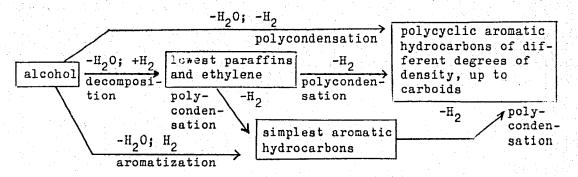
alcohol catalytic olefins olefins catalytic dehydro-genation dehydro-genation

polycyclic saturated, unsaturated, and aromatic hydrocarbons of different density degrees, up to carboids

The replacement of the intermediate mechanism by the high-temperature mechanism manifests itself by a marked retardation of the process at temperatures above 750°C. In the range of 750° - 875° the carbon formation is slowed down, after which it is again speeded up vigorously beyond 875°C. This mechanism (750° - 950°C) comprises the following processes: polycondensation of decomposition products of the alcohols used (lowest paraffins and ethylene); polycondensation of aromatic hydrocarbons, resulting from the decomposition of alcohols; polycondensation of the alcohols used. A general scheme applies to them:

Card 3/5

Carbon Formation in the Decomposition of \$\ \frac{\\$5/062/60/000/011/003/016}{\}\$ Isopropyl Alcohol, n-Hexyl Alcohol, and \$\ \text{B013/B078}\$ Cyclohexanol on the Copper - Silica Gel Catalyst



As opposed to the two first-mentioned mechanisms, aromatic hydrocarbons only are given here as the end products. This is explained by the fact that under the conditions of the high-temperature mechanism the formation of saturated, unsaturated, and hydroaromatic polycyclic systems is practically impossible, which is indicated by the composition of

Card 4/5

resinous polycondensation products. Their composition and aromatic character is almost the same in all of the alcohols investigated. Table 1 gives the composition of gaseous decomposition products of the alcohols on the copper - silica gel catalyst. The composition of the hydrocarbon part of the gaseous decomposition products of isopropyl and n-hexyl alcohols on the copper - silica gel catalyst is given in Table 2. There are 4 figures, 2 tables, and 4 references: 3 Soviet and 1 German.

ASSOCIATION: Moskovskiy gosudarstvennyy universitet im. M.V. Lomonosova

(Moscow State University imeni M.V. Lomonosov)

SUBMITTED: June 29, 1959

Card 5/5

RUDENKO, A.P.; BALANDIN, A.A.; ZABOLOTNAYA, M.M.

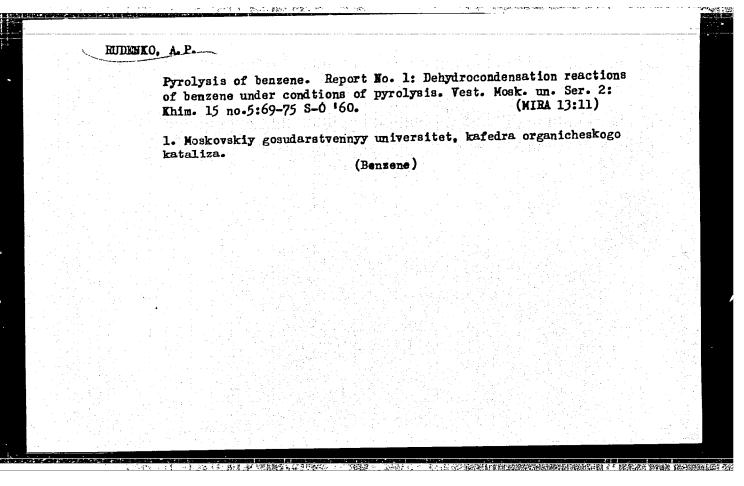
Mechanism of coal formation during the decomposition of methane, ethane, ethylene, and acetylene on silica gel. Izv.AN SSSR.Otd.khim. nauk no.6:989-995 Je '61. (MIRA 14:6)

1. Moskovskiy gosudarstvennyy universitet im. M.V.Lomonosova. (Hydrocarbons) (Coal)

STEGNER, G.; RUDENKO, A.F.; BALANDIN, A.A.

Carbon formation during the decomposition of isopropyl alcohol, n-hexyl alcohol. and cyclohexanol on a copper-silica gel catalyst. Izv. AN SSSR.Otd. khim. nauk no.11:1930-1937 N '60. (MIRA 13:11)

1. Moskovskiy gosudarstvennyy universitet im. M.V.Lomonosova. (Isopropyl alcohol) (Hexanol) (Cyclohexanol)



khim. nauk nc. 1:164-166 Ja '61. 1. Moskovskiy gosudarstvennyy universitet im. M.V. Lomonosova. (Hydrocarbons) (Carton)			Factors causing a change in the mechanisms of carton formation during the decomposition of hydrocarbons. Izv. AN SSSR. Otc.	
(Hydrocarbons) (Carbon)			khim. nauk nc. 1:164-166 Ja !61. (MIRA 14:2)	
			1. Moskovskiy gosudarstvennyy universitet im. M.V. Lomonosova.	
			그 이 네는 이 사람들이 모일을 그리지 않는 이번 가능하다면 어떻게 된다.	
				•
				5.5
			이번 사람들은 사람이 되었다. 그는 그는 그 그리고 한 사람들이 모르는 수 없었다.	
				11/2
			"大","大","我们就是"大","大","大","我们就是一个人,我们就是一个人,我们就是一个人,我们就是一个人,我们就是一个人,我们就是一个人,我们就是一个人	
		21.	보는 이 문문은 일 하는 한 생각을 들어 그 위한 이 아내는 이 에 대한 생각을 하는 것 같아 한 것 같아? 작품을 하는	
			그는 얼마 그리고 아내가 되는 사람들이 가장 하는 것이 되는 것이 되는 것이 없는 것이 없었다. 그 사람들은 하는 것이 없는 것이 없는 것이 없다면 하는 것이다.	
			그가 있는데 남자가 되는 그 다른 그리다 나는 그런 얼마는 사람들이 그리는데 되고 되다는데 하네요.	1
			在这个是一个一点,是是这个人,就是一定的一点的一个大人,就就是这个意思的能够是一个。	
				4.47
			그 그 전 그는 일이 되는 그 학생들이 없었다. 그렇게 그렇게 그 그는 그는 하실 생각을 통하고 한 점계되어 되다.	9.0
ĸĸĸĸĸĸĸĸĸĸĸĸĸĸĸĸĸĸĸĸĸĸĸĸĸĸĸĸĸĸĸĸĸĸĸĸĸ	1.00		그는 그는 어느 그는 그들은 전문에 들었다면 그 아이가 아니라 나를 마음하는 그리고 됐다면 그 때에게	
도 마양된 사람들은 사람들이 되는 것이 되었다. 그렇게 되는 것이 되었다. 이 사람들은 사람들이 얼마를 하는 것이 되었다. 그는 것이 그렇게 되었다. 그렇게 되었다. 그렇게 되는 것이 되었다. 사람들은 사람들은 사람들은 사람들은 사람들은 사람들은 사람들은 것이 되었다.			그는 한 문 이 나는 그를 가는 것이 없는 얼마를 되었다. 이 그 그 것은 이 그는 속을 하고 불을 받았다. 그 그 그	
그는 분들이 되었다. 그는 그는 그는 그는 그를 보고 하는데 보고 있다. 그는 그리고 보이를 보고 있다. 그는 그를 보고 있는데 보고 있다. 그는 것은 것은 그는 그는 그는 그는 그는 그는 그는 그들은 것은 그를 보고 있는데 그를 보고 있는데 그를 보고 있다.	1.35		自己,1986年,1987年,教育的基本的基本的特別,在中国国际,就是1986年,和基本和基本的特殊的第二人员	1.5
그들로 맞았는 그렇지요 그는 마른 그림을 다고 먹었다. 얼굴 하고 말라는 얼굴을 다 하고 있다. 그 그 그 그 그 그 그 그 그 그 그 그 그 그 그 그 그 그 그		eggi (1966) etter George	그리네는 그리고 한 전염 다양하는 학생들은 하지만 하는 일반에 되었다. 그렇게 나를 하는 하는 하는 것은 하는 것이다.	2.1
,这个时间,她们就是一点点,就是一个女子,只是一个人,我们就是一个人,我们就是一个人,我们就是一个人,我们就是一个人,我们就是一个人,我们就是一个人,这个人,不			그리는 이 이용 그 하는 생활님 하다면 나고 보고 보지 않는 아래를 받는 수 없는 것이 없는 데 보다 하다.	100
,一个一个一个一个一个一个一个一个一个一个一个一个一个一个一个一个一个一个一个			님은 내고 그 도움을 통한 되어 되게 되었다면서 관심하다면서 달라지 않는 사이를 느꼈다면 나는 그는	

s/079/60/030/009/021/022/XX B001/B066

11.4600

Rudenko, A. P. and Dobrosel skaya, N. P. AUTHORS:

TITLE:

The Role of Complexing Additions in the Synthesis of Phthalocyanines 11. Action of Phosphoric Acid Compounds and Molybdenum Trioxide on the Formation Reaction of

Fe Phthalocyanine

PERIODICAL:

Zhurnal obshchey khimii, 1960. Vol. 30. No. 9,

pp. 3077 - 3083

The formation of phthalocyanine and its metallic compounds from phthalic acid derivatives can, according to published data, be accelerated catalytically by adding different products, i. e., metals, metallic oxides, chlorides, sulfates, and other compounds of the elements of groups I-VI. The compounds of groups V-IV of the periodic system are known to be particularly active catalysts. It was the objective of the present paper to study the catalytic effect of the most active of these catalysts, and to explain the formation mechanism

Card 1/3

The Role of Complexing Additions in the S/079/60/030/009/021/022/XX Synthesis of Phthalocyanines I. Action B00:/B066 of Phosphoric Acid Compounds and Molybdenum Trioxide on the Formation Reaction of Fe Phthalocyanine

of phthalocyanine. The influence of phosphoric acid compounds and molybdenum trioxide upon the yield of Fe phthalocyanine (II) obtained by reaction of iron dust with melted phthalimide (I) or phthalamide (III) in ammonia at 240° and standard pressure was studied. Addition of 0,1 mole molybdenum and phosphorus in the form of crthophosphoric acid, monosubstituted ammonium phosphate, molybdenum trioxide, or ammonium mclybdate lowers the formation rate of Fe phthalocyanine (II) from phthalamide (III) and iron. The same admixtures of heteropoly compounds, such as phosphomolybdic acid and ammonium phosphomolybdate, increase the reaction rate. The catalytic effect of the admixtures becomes manifest in the condensation of phthalamide (formation of new C-N bonds) The effect of admixtures on the condensation of phthalamide is found to depend on their capability of adding and splitting off ammonia as well as of forming and splitting the complexes 7 with the initial phthalamide and its condensation products. The activating effect of admixtures is observable during the formation of labile

Card 2/3

The Role of Complexing Additions in the Synthesis of Phthalocyanines. J. Action of Phosphoric Acid Compounds

S/079/60/030/009/021/022/XX B001/B066

and Melybdenum Trioxide on the Formation Reaction of Fe Phthalocyanine

complexes. During the formation of stable complexes, like in the case of phosphoric acid, a passive behavior was noted when adding 1 mole of acid per : mole of phthalamide until the process was completed. On the basis of the experimental results, a formation mechanism of Fe phthalocyanine from phthalamide and iron is suggested, which represents a multi-stage polycondensation of phthalamide. The effect of small or large admixtures on the yield of Fe phthalocyanine is illustrated in three diagrams. There are 3 figures. 1 table, and 30 references: 6 Soviet, 4 US, 7 British, 12 German, and 1 French.

ASSOCIATION:

Moskovskiy gosudarstvennyy universitet

(Moscow State University)

SUBMITTED:

August 6, 1959

Card 3/3

RUDENKO, A.P.; DOBROSELISKAYA, N.P.

Role played by complexing addition agents in the synthesis of phthalocyanines. Part 2: Effect of the compounds of phosphoric acid, chromium oxide, and tungsten trioxide on the formation of Fe-phthalocyanine. Zhur. ob. khim. 31 no. 11:3667-3671 N '61. (MIRA 14:11)

1. Moskovskiy gosudarstvennyy universitet imeni M.V. Lomonosova. (Phthalocyanine) (Complex compounds)

STANISHEVSKIY, A.S.; RUDENKO, A.P.; YAGODIN, A.N.

Methods for calculating a heavy drill-stem bottom. Trudy
VITR no.3:39-69 161. (MIRA 15:7)

(Boring machinery)

RUDENKO, A.P.

Effective penetration per a core drilling cycle . Izv. Tys.ucheb.zav.;
geol.i razv. 5 no.6:121-125 Je '62. 'MIRA 15:7)

1. Vsssoyuznyy nauchno-issledovatel'skiy institut tekhniki razvedki.
(Boring)

(MIRA 14:2)

Pyrolysis of benzene. Report No.2: Characteristics of the pyrolysis kinetics of benzene. Vest.Mosk. un. Ser. 2: Khim. 15 no.6:66-71 N-D

160.

1. Kafedra organicheskogo kataliza Moskovskogo universiteta. (Benzene)

18.7200

77734 sov/149-60-1-23/27

AUTHOR:

Rudenko, A. P.

TITLE:

Concerning Stabilization of Lithium Fluxes for

Welding Aluminum and Its Alloys

PERIODICAL:

Izvestiya vysshikh uchebnykh zavedeniy.

metallurgiya, 1960, Nr 1, pp 150-153 (USSR)

ABSTRACT:

Active ingredients in aluminum welding fluxes are metal fluorides which dissolve aluminum oxide film and achieve a close contact of the welding material with aluminum. Chlorides in fluxes depress mp but fail to cause deoxidation. In previously published

works it was generally accepted that lithium chloride is a requisite of high flowability in flux.

The following fluxes were investigated:

Card 1/3

Concerning Stabilization of Lithium Fluxes for Welding Aluminum and Its Alloys

7773⁴ SOV/149-60-1-23/27

Table 1: Composition of fluxes, %.

FLUX Nr Couponents	1	Z	3 (AF-4a)	4	5
KCI NaCI LICI KP NaF LIF	50 5.5 ,0 3 1.5	30 43 10 15	50 98 14 8	48 46 — 1.1 4.9	26,5 30 11,8 —

Fluxes containing LiCl are deliquescent causing chemical and physical reactions: chloride recrystallization accompanied by the formation of lumpy aggregates, and interaction of LiCl and KF with the formation of insoluble LiF. Homogeneity of the flux is disrupted affecting the weld

Card 2/3

Concerning Stabilization of Lithium Fluxes for Welding Aluminum and Its Alloys

77734 SOV/149-60-1-23/27

quality adversely. The flux can be regenerated by grinding proving that the new crystal formation and loss of homogeneity rather than the formation of lithium fluoride cause weld inadequacy. Exclusion of hygroscopic components (KF and LiCl) and substitution of LiF for LiCl is the logical conclusion of these observations. This was experimentally proved by excellent welding results with flux Nr 4, superior to all others and not hygroscopic. Further study revealed that the elimination of oxide films on aluminum is exclusively caused by fluorides, mainly LiF, the presence of which is more essential than that of LiCl. There are 2 tables; and 9 Soviet references.

ASSOCIATION:

Moscow State University. Chemical Department. Chair of Organic Catalysis (Moskovskiy gosudarstvennyy universitet. Khimicheskiy Fakultet. Kafedra organicheskogo kataliza)

SUPNITTED:

May 4, 1959

RUDENKO, A.P.; RCDICHEVA, M.F.; LEONT'YEV, Ye.A.; LUKINA, T.V. (Moscow)
"Macromechanism" of carbon formation in the decomposition of

"Macromechanism" of carbon formation in the decomposition of benzene on compressed carbon black. Zhur. fiz. khim. 38 no.3: 616-622 Mr '64. (MIRA 17:7)

1. Moskovskiy gosudarstvennyy universitet imeni M.V. Lomonosova.

"APPROVED FOR RELEASE: 06/20/2000

CIA-RDP86-00513R001445920001-5

<u>L 25156-65</u> EWT (m)/EPF(c)/EWP(j) Pc-4/Pr-4 ACCESSION NR: AP5002000

-4 RM

5/0020/64/159/006/1374/1377

AUTHOR: Rudenko, A. P.

TITLE: Self-developing catalytic systems

SOURCE: AN SSSR. Doklady, v. 159, no. 6, 1964, 1374-1377

TOPIC TAGS: self developing catalytic system, evolutionary catalysis, catalysis property, catalytic process

ABSTRACT: The author is of the opinion that the study of the evolutionary regularities of the catalytic processes may form a new branch of catalytic chemistry, the evolutionary catalysis, in which the variable nature of the catalyst is taken into consideration. A number of papers are quoted in which the changes of the catalysts during the reaction is noted. The accidental changes in the nature of the catalyst may take place as a result of interactions with the products of the reactions and of microfluctuations, and the nature of the catalytic centers may gradually change. The changes might be in the crystalline structure, in the ab-

Card 1/2

L 25156-65 ACCESSION NR: AP5002000			
sorbing, physical and chemi in terms of the probability t equations.	neory. Orig. at		
ASSOCIATION: Moscovskiy (Moscow State University)	gosudarstvennyy univ	ersitet im. M. V. Lomo	nosova
SUBMITTED: 23Jun64	ENCL: 00	SUB CODE: GC	
NR REF SOV: 008	OTHER: 001		
Card 2/2			

CIA-RDP86-00513R001445920001-5 "APPROVED FOR RELEASE: 06/20/2000

507/62-60-1-5/37 5.3000

Stegner, G., Balandin, A. A., Rudenko, A. P.

AUTHORS: Influence of Different Stages of Polycondensation of the Products of Catalytic Decomposition of Ethyl TITLE:

Alcohol on the Rate of Carbonization

Izvestiya Akademii nauk SSSR. Otdeleniye khimicheskikh

nank, 1960, Nr 1, pp 24-30 (USSR) PERLODICAL:

This is a continuation of the author's previous work (Izv. AN SSSR, Chem. Ed., 1959, 1896) on the mechanism ABSTRACT:

of carbonization in the decomposition of ethyl alcohol over copper-silica. Experimental data presented in this paper confirm previous conclusions (see above

reference) concerning the mechanism of carbonization which accompanies catalytic decomposition of ethyl alcohol. Carbonization is considered to be a multistage polycondensation of ethyl alconol and the products of its catalytic decomposition. The so-called

low temperature carbonization (below 600°) proceeds

Card 1/2

Influence of Different Stages of Polycondensation of the Products of Catalytic Decomposition of Ethyl Alcohol on the Rate of Carbonization

78059 807/62-60-1-5/37

through dehydrogenation of ethyl alcohol. Acceleration or slowing down of ethyl alcohol decomposition (dehydrogenation and dehydration) causes a change in the rate of carbonization. There are 4 figures; and 11 references, 1 German, 10 Soviet.

ASSOCIATION:

M. V. Lomonosov Moseow State University (Moskovskiy gosudarstvennyy universitet imeni M. V. Lomonosova)

SUBMITTED:

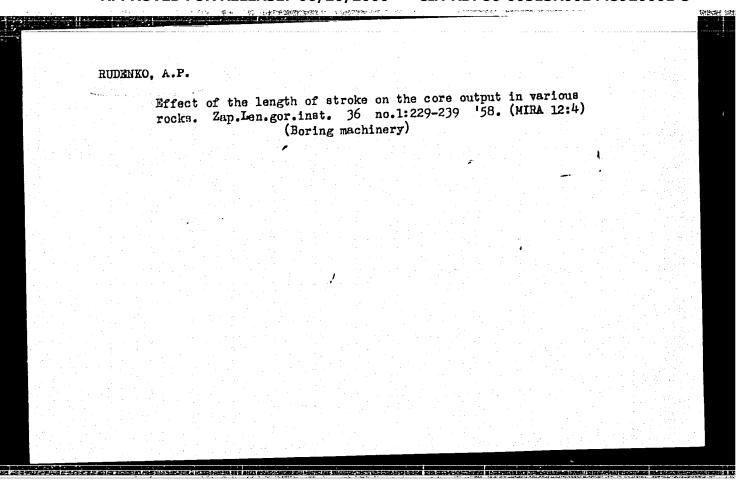
May 4, 1958

Card 2/2

AGRONOMOV, A.Ye.; PATRIKEYEV, V.V.; RUDENKO, A.P.

Nonhomogeneity of the structure of silica gel. Vest. Hosk.un.
Ser.mat., mekh., astron., fiz.khim. 13 no.3:197-206 58.
(HIRA 12:4)

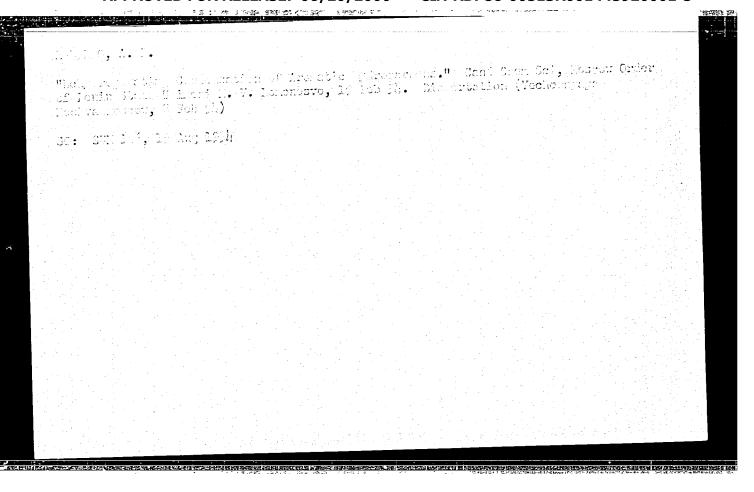
1. Kafedra organicheskogo kataliza Moskovskogo universiteta.
(Silica)



RUDENKO, A.P.; BALANDIN, A.A.; KACHAN, S.Ya.

Two mechanisms of carbon formation in the course of the decomposition on silica gel, of n-paraffins, naphthenes, and aromatic hydrocarbons having six and seven carbon atoms. Izv.AN SSSR.Otd.khim.nauk no.6:981-988 J1 '60. (MIRA 13:7)

1. Moskovskiy gosudarstvennyy universitet imeni M.V. Lomonosova. (Paraffins) (Naphthenes) (Pyrolysis)



l. Leningradskiy gornyy institut. (Oil well drilling fluids) (Borings)	ucheb. zav.; nert	g fluids on the core in a i gaz no.4:45-49 158.	2 0010 00110-1	(MIRA 11:9)	
	1.Leningradskiy go	ornyy institut. (Oil well drilling flu	ids) (Borings)		
		trator (Sept. 1993). Primary and section (Sept. 1997).			

CIA-RDP86-00513R001445920001-5 "APPROVED FOR RELEASE: 06/20/2000

5(3)

Agranomov, A.Ye., Patrikeyev, V.V.

sov/55-58-3-24/30

AUTHORS:

TITLE:

and Rudenko, A.P.

On the Inhomogeneity of the Structure of Silica Gel (0

neodnorodnosti struktury silikagelya)

PERIODICAL:

Vestnik Moskovskogo universiteta, Seriya matematika, mekinaika,

astronomii, fiziki, kiidii ,1958, Nr 3, pp 197-206 (USSR)

ABSTRACT:

The silica gel ASK of the Chemical Combinate in Voskresensk was investigated. The structure is inhomogeneous inasmuch as different single piaces absorb differently strongly the phthalocyanin of copper from a solution. Using the color differences the authors obtained test pieces with homogeneous structure in mechanical way. It was stated that only those test pieces are able to absorb the phthalocyanin, the pore entrances of which are at least twice as great as the molecules of the coloring substance. Furthermore: the inhomogeneity originates by mixture of three different structures with dense particle packing and of several intermediate structures. The results of A.V. Kiselev, G.K. Boreskov, I.Ye. Neymark, R.Yu. Sheynfayn, and others are used.

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20

On the Inhomogeneity of the Structure of Silica Gel SOV/55-58-3-24/30

There are 5 figures, 4 tables, and 21 references, 9 of which are Soviet, 9 English, 2 American, and 1 German.

ASSOCIATION: Kafedra organicheskogo kataliza (Chair of Organic Catalysis)

SUBMITTED: June 17, 1957

Card 2/2

RUDENKO, A. P. Master Tech Sci (diss) -- "The effect of the action of wash fluid on the completeness with which a core is obtained". Leningrad, 1958.

15 pp (Min Higher Educ USSR, Leningrad Order of Lenin and Order of Labor Red Banner Mining Inst im G. V. Plekhanov, Chair of the Development of Oil and Gas Deposits), 110 copies (KL, No 7, 1959, 126)

5 (3) AUTHORS: Rudenko, A. P., Kazanskiy, B. A.,

SOV/20-128-1-26/58

Academician

TITLE:

Heterogeneous-catalytic Course of Benzene Pyrolysis Reactions

PERIODICAL:

Doklady Akademii nauk SSSR, 1959, Vol 128, Nr 1, pp 99-102 (USSR)

ABSTRACT:

The present paper investigates the role played by the contact surface during benzene pyrolysis especially in the formation of diphenyl as well as the specificity of different contacts. The enlargement of the contact surface had positive results and led to the conclusion that the reaction is of heterogeneous catalytic character (Table 1). In the investigation of the specificity of individual catalysts a number of applied catalysts - such as metal on silica gel - were employed (Table 2). The specificity of the catalysts investigated indicates a marked distinction (Figures 1 and 2). On the basis of the results obtained the kind of contact during pyrolysis is not all unimportant as was sometimes assumed (Refs 4 and 5). It determines the marked peculiarities of the specificity of contacts and their catalytic activity with respect to individual reactions of the dehydrocondensation of benzene. The manifestation of the specificity of contacts speaks also for a heterogeneous catalytic

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Heterogeneous catalytic Course of Benzene Pyrolysis SOV/20

SOV/20-128-1-26/58

mechanism of the benzene pyrolysis. The physical changes of contact surfaces must also be pointed out; they only take place in the reaction zone and cannot be observed in an inert medium (nitrogen) on simple heating. It may be assumed that due to the heterogeneous catalytic character of the diphenyl formation the migration of atoms and atomic groups of the centact is facilitated by the formation of catalytic complexes. There are 2 figures, 2 tables, and 14 references, 7 of which are Soviet.

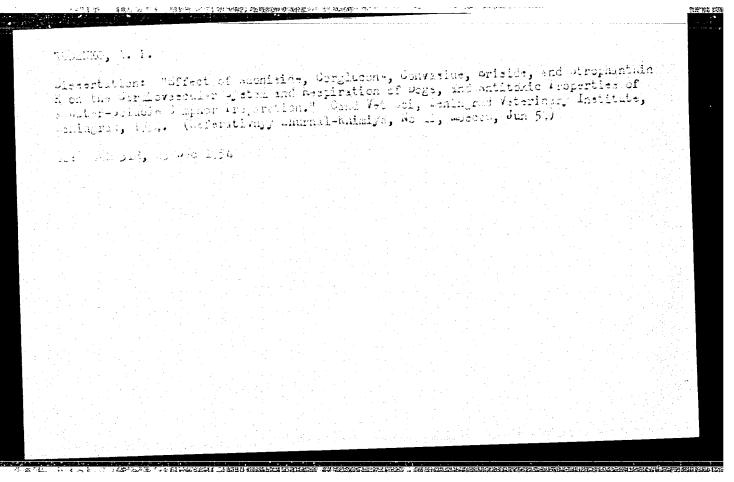
ASSOCIATION:

Moskovskiy gosudarstvennyy universitet im. M. V. Lomenosova (Moscow State University imeni M. V. Lomenosov)

SUBMITTED:

May 26, 1959

Card 2/2



BALANDIN, A.A.; RUDENKO, A.P.; STEGNER, G.

Formation of coal dendrites in the course of decomposition of alcohols on nickel. Ozv.AN SSSR.Otd.khim.nauk no.5:762-770 My 161.

l. Moskovskiy gosudarstvennyy universitet im. M.V.Lomonosova.

(Alcohols) (Coal)

Bal/NOIN. A.A.; PATRIXYPV, V.V., SHAKNOVA, P.K.; PHDEMSO, A.I. (Fracew)

Setermination of unemical additional by the differential thermocouple method. Whire fiz. knim. 36 no.9:1952-1957 S '62.

(M.R. 17:6)

1. Institut organicheskov knimil al 1905.

Con the time and apparent activities of heterogeneous catalysts.

Vent. Mosk. un. Ser. 2: Khim. 18 no.5:57-61 S=0 '63.

(MIRA 16:11)

1. Hafefra organicheskogo kataliwa Moskovskogo universiteta.

RUDENKO, A.P.; BODRINA, D.E.; BALANDIN, A.A., akademik; RODICHEVA, M.F.

Alkylation of benzene by a coaly substance obtained from propylene on silica gel. Dokl. AN SSSR 165 no.4:874-877 D 165. (MIRA 18:12)

1. Moskovskiy gosudarstvennyy universitet im. M.V. Lomonosova.

RIDERKO, A. S.

Academy of Sciences - Geologists Sep 50

"New Problems of Genetic dineralogy" Prof D. P. Grigor yev, Priroda No 9, pp 22-30

Mentions the following persons as contributing greatly to the development of the science in the USSR: G. G. Lemmleyn, Leningrad/Moscow; I. I. Shafranovskiy, Leningrad; G. N. Vertushkov. (information incomplete)

RUDENKO, A.T., kandidat meditsinskikh nauk (Leningrad)

Gourse of open fractures of the mandible in radiation sickness;
experimental study. Stomatologiia 36 no.2:31-33 Mr-Ap 157.

(MLRA 10:6)

(RADIATION SIGKNESS) (JAW--FRACTURE)

RUDENKO, A.T., kandidat meditsinskikh nauk

Pathological changes in the oral cavity in radiation sickness.
Stomatologiia 35 no.4:7-11 J1-Ag '56 (MLRA 10:4)

1. Iz Voyenno-morskoy meditsinskoy akademii (RADIATION SICKNESS) (MOUTH--DISEASES)

RUDENKO, Anatoliy Terent'yevich; UVAROV, V.M., red.; KCNONOVA. L.B., tekhn. red.; CHUNAYEVA, Z.V., tekhn. red.

[Pathology of the dentition of wisdom teeth]Fatologiia prorezyvaniia zubov mudrosti. Leningrad, Medgiz, 1961. 63 p. (MIRA 15:3)

"The Course of Open Fractures of Mandible During Radiation Sickness," (Experimental Research), By A. T. Rudenko, Candidate of Medical Sciences, Leningrad, Stomatologiya, No 2, Mar/Apr 57, pp 31-33

The present research traces certain general mechanisms in the course of jaw and facial traumas during radiation sickness.

Tests were conducted on 21 rabbits which were classified into seven groups of three rabbits each. In each group one rabbit was subjected to both trauma of the middle third of the mandible plus irradiation by 700 r, one rabbit was subjected to trauma only and one rabbit was irradiated by 700 r only.

Symptoms of acute radiation unjury appeared significantly earlier (third to fourth day) in animals subjected to both trauma and irradiation than in those subjected to irradiation only (seventh to tenth day).

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The duration of the life of animals subjected simultaneously to trauma and irradiation was significantly shorter than the duration of the life of animals subjected to trauma only.

Necrotic processes and tissue decomposition were distinctly predominant over reparative processes when traumas of the jaw and facial regions were combined with radiation sickness.

Active tissue inflammation, as a protective reaction of the organism, was depressed when trauma was combined with radiation sickness. (U)

Sum 110 1451

USSR / Human and Animal Morphology, Normal and Pathological.

Digestive System.

: Ref Zhur - Biol., No 8, 1958, No 35923 Abs Jour

: Rudenko, A. T. Author

: Not given : Pathological Changes of the Oral Cavity in the Radiation Inst Title

Disease.

: Stonatologiya, 1956, No. 4, 7-11 Orig Pub

: Rabbits were subjected to single general X-ray exposures in doses of 700 (I), 1000 (II) and 2,000 (III) r. All Abstract

animals contracted the radiation disease, the most serious one taking place during III. After I, changes in the mucous membrane were not conspicuous. After II, in some cases, there occurred hemorrhages of the nucous membrane and also ulcerations on the lingual dorsum and in the area of the tonsils. After III, extensive ulcerously-negrotic

Card 1/2

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Briuks, V.R.; Rub The, A.L.; Rubayavisav, K.L.
Investigation of one black daving processes. Shor. nauch. trud.
RCRI no.232 43-52 *53 (AIRA 17:8)

LIESTER, B.N.; RUDENKO, A.T.

Case of a Barre-Masson tumor. Ortop., travm. i protez. 26 no.12:66 D 165.

(MIRA 19:1)

1. Iz ortopedo-travmatologicheskogo otdeleniya (zav. - B.N. Libster) 2-y gorodskoy bol'nitsy Luganska (glavnyy vrach - A.T. Chumakova). Adres avtorov: Lugansk, ul.Frunze, d.106, 2-ya gorodskaya bil'nitsa. Submitted March 17, 1965.

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